Iowa Science Teachers Journal

Volume 14 | Number 1

Article 6

1977

A Solar Water Heater

John Bridgenell Solar Energy Society of America

Follow this and additional works at: https://scholarworks.uni.edu/istj

Part of the Science and Mathematics Education Commons

Let us know how access to this document benefits you

Copyright © Copyright 1977 by the Iowa Academy of Science

Recommended Citation

Bridgenell, John (1977) "A Solar Water Heater," *Iowa Science Teachers Journal*: Vol. 14: No. 1, Article 6. Available at: https://scholarworks.uni.edu/istj/vol14/iss1/6

This Article is brought to you for free and open access by the IAS Journals & Newsletters at UNI ScholarWorks. It has been accepted for inclusion in Iowa Science Teachers Journal by an authorized editor of UNI ScholarWorks. For more information, please contact scholarworks@uni.edu.

Offensive Materials Statement: Materials located in UNI ScholarWorks come from a broad range of sources and time periods. Some of these materials may contain offensive stereotypes, ideas, visuals, or language.

A SOLAR WATER HEATER

John Bridgenell Solar Energy Society of America Torrance, California 90510

The sketch (Fig. 1) and photograph (Fig. 2) show a very simple solar water heater made from readily available parts. The heater was constructed by myself, with a little help from Dad, for my science fair project. Incidentally, it won first prize in the seventh grade physics division.



Fig. 2. A solar water heater.

Fig. 1.

A solar water heater.



The heater uses the thermosyphon principle to accumulate hot water in the tank and to return the cooler water to the solar collector panel. For this reason, the water tank (a 3-lb. coffee can) is located above the solar panel. The panel is made of 3/16-inch diameter copper tubing soldered to a $12\frac{1}{2}$ by 20-inch zinc. sheet at about $2\frac{1}{2}$ inch intervals. The 3/16-inch tubes are manifolded into $\frac{1}{2}$ -inch diameter copper tubing at the top and bottom of the panel. The panel is enclosed in a wooden tray with a plastic sheet lid. The water pipes are connected to the tank by transparent plastic tubing.

The panel was positioned at about a 34-degree angle to the horizontal (our latitude is about 34 degrees) and pointed about 10 degrees west of south. The temperature rise of water in the tank was plotted versus the time of day with an unpainted panel, a painted panel, and with a double layer of plastic sheet over the panel. The depth of the thermometer in the water tank could be varied so that the maximum and mean water temperature could be measured. Painting the panel with a flat black paint raised the maximum water temperature by about 25%, while using a double layer of plastic sheet increased the temperature an additional 2%. The maximum water temperature obtained was 178 degrees F with the water starting out at 65 degrees in the morning. The maximum ambient air temperature on this day was 66 degrees F.

The maximum efficiency of the heater was calculated by measuring the mean water temperature rise per hour and comparing the heat required to achieve this (in Btu per hour) to the incident heat from the sun reaching the solar panel. It was assumed that 50% of the 420 Btu/hr/ft² incident solar radiation outside the atmosphere reached the panel. On this basis the highest efficiency was obtained with the panel painted flat black and with a double layer of plastic. The maximum efficiency was calculated to be 45% and occurred at 10:45 AM when the mean water temperature was 105 degrees F. As the water temperature got higher, the efficiency fell off due to heat loss and re-radiation from the panel. At 3 PM, when the water temperature reached a maximum, the overall heater efficiency was zero. After this time, the water started to cool. It was also shown that the cooling rate could be delayed slightly by covering the panel at night.

*Reprinted by permission of the Solar Energy Society of America from its journal *Energies*. The journal has an education section with many ideas about solar energy projects.

Energy

If energy was delivered to your door like milk-today, you would have had 19 half gallons of oil, 14 half gallons of natural gas, and 46 pounds of coal on your doorstep. By 1985, your family will need 30 half gallons of oil, 24 half gallons of natural gas and 70 pounds of coal. The demand for energy throughout the world is rising fast. We must not waste.....but use it wisely.

Environmental Vistas 8(3):2